





University of Durham



A Thesis Entitled

**Novel Conjugated Schiff-Base Compounds**

Submitted by

**Wan Mohd Khairul Wan Mohamed Zin  
BSc. Honours (UPMalaysia)**

Ustinov College

Department of Chemistry  
University of Durham,  
University Science Laboratories,  
South Road, Durham DH1 3LE, UK

A Candidate for the Degree of Master of Science  
2004

*To Faith and War*

*for love, courage and truth...*



**Ustinov College**  
**University of Durham**

## Acknowledgement

*To Mek and Wae*

*for love, courage and trust...*

My first deepest gratitude is of course to Dr Peter Lam for his kind supervision, help and guidance during the course of my MSc. To be honest, without his support I could never have done this far. I would also like to thank Dr Donpeth Lydon for his true friendship, motivation and support for the past year, although sometimes he had to be a bit harsh with his harsh thoughts, but it is all worth it at the end, without him I also think that I would not complete my thesis. Thanks are also extended to Prof Todd Marder for his technical and theoretical supports for the past year of my studies. Only God can pay all the good debts of what all of you have done. From the depths of my heart, I would not come this far without continuous blessing, support and prayers from my beloved parents, Wan Muhamad Zin Wan Adyad and Rokiah Awang as well as my family, many thanks are given to all of them.

Thanks also goes to the BRD (R & T) Division, Ministry of Science, Technology and Innovation Malaysia and the staff of University College of Science and Technology Malaysia, without their support and help, it is impossible for me to do the research and complete this research.

Special thanks also given to Rachel Roberts, Richard Gordon, and Ibrahim Adnan for friendship, help, support and definitely for laughter for all these days.



## Acknowledgement

My first deepest gratitude is of course to Dr Paul Low for his kind supervision, help, and guidance during the course of my MSc. To be honest, without his support I would never come this far. I would also like to thank Dr Donocadh Lydon for his true friendship, motivation and support for the past year, although sometimes it's hard for me to take with his harsh thought, but it is all worth it at the end, without him I also think that I would not complete my thesis. Thanks are also extended to Prof. Todd Marder for his technical and theoretical supports for the past year of my studies. Only God can pay all the good deeds of what all of you have done. From the deepest of my heart, I would not come this far without continuous bless, support and prayers from my beloved parents, Wan Mohamed Zin Wan Ahmed and Rokiah Awang as well as my family, many thanks are given to all of them.

Thanks also goes to the HRD (S & T) Division, Ministry of Science, Technology and Environment Malaysia and the team of University College of Science and Technology Malaysia, without their support and help, it is impossible for me to be here and complete this research.

Special thanks also given to Rachel Roberts, Richard Cordiner, and Ibraheem Mkhaliid for friendship, help, support and definitely for laughter for all these days.

I would also to thank Dr Andy Beeby, Dr Graham Sandford, Dr Andrew Hughes, Dr Ritu Katakya, Dr Tolu Fasina, Dr Stephane Gaillard and Dr Jon Collings for their theoretical and practical supports during my studies.

Thanks also extended to the department technical staff, namely: Dr Alan Kenwright, Mrs. Catherine Hefferman and Mr Ian McKeag (NMR); Dr Mike Jones and Miss Lara Turner (Mass Spectroscopy); Mrs Jaraka Dostal (Elemental Analysis); Mr Peter Coyne and Mr Malcolm Richardson (Glass Blowing); Ms Judith Magee, Mrs Emma Dowson and Mr Brian Eddy (Undergraduate Teaching Lab); Mr Barry Baker and Mr Kelvin Appleby (Electrical Technicians); Mr David Hunter (High Pressure Operations); Mr Douglas Carswell (DSC works); Mr Jimmy Lincoln, Mrs Elizabeth Wood, Mr Anthony Baxter (Stores) and Dr Euan Ross, Dr Hillary Hull and Mrs Jean Eccleston (Administration), without the aid and support from all of them this research would not have been possible. I also thank all the other people who worked and studied in the chemistry department who have not been mentioned.

To all Lab 007 members, namely; Dr Laurent Poress, Dr Aileen Congreve, Simon Rutter and last but not least Karen Findlay for their support and patience for guiding me until the end of my research (spectroscopic studies).

Finally to all past and present friends (you know who you are), thanks for all the friendship, trust, support and laughter for the past years, It is such a golden opportunity to get to know all of you.

### Statement of Copyright

No part of this thesis may be reproduced by any means, nor transmitted, nor translated into any other form or language without the written permission of the author.



## Memorandum

The work presented within this thesis was carried out in the University of Durham between October 2003 and September 2004. This thesis is the work of the author, except where acknowledged by reference and has not been submitted for any other degree. The copyright of this thesis lies solely with the author and no quotation from it should be published without prior written consent and information derived from it should be acknowledged.

## Statement of Copyright

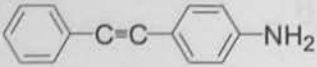
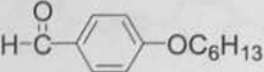
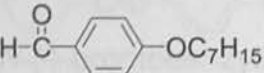
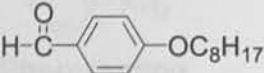
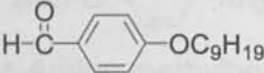
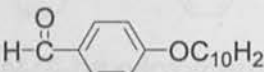
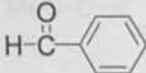
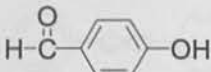
No part of this thesis may be reproduced by any means, nor transmitted, nor translated into any machine or language without the written permission of the author.

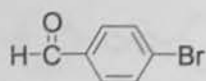
## Abbreviations

$\lambda$	Wavelength
$\epsilon$	Extinction coefficient
A	Acceptor
Cr	Crystalline phase
CTAB	Cetyltrimethylammonium bromide
D	Donor
DCM	Dichloromethane
DIPEA	Diisopropylethylamine
DME	Dimethoxyethane
DMF	Dimethylformamide
DSC	Differential scanning calorimetry
DTA	Differential thermal analysis
Ed.	Edited
EI - MS	Electron Ionisation - Mass Spectrometry
Equiv	Equivalent
HOMO	Highest occupied molecular orbital
Hz	Hertz
I	Isotropic phase
IR	Infrared
LBTs	Langmuir-Blodgett Techniques
LCs	Liquid Crystals
LUMO	Lowest unoccupied molecular orbital

Max	Maximum	
mM	milimolar	
n	Vector quantity (director)	16a
N	Nematic phase	17
nm	Nanometre	
NMR	Nuclear Magnetic Resonance	
OLEDs	Organic Light-Emitting Diodes	2a
pKa	Acid dissociation constant	
PPEs	Poly(aryleneethynylene)s	2b
R	Alkyl	2b
RT	Room temperature	
SAMs	Self-Assembled Monolayers	2c
SmA	Smectic A phase	
SmB	Smectic B (hexatic B) phase	2d
SmC	Smectic C phase	2d
SN	Substitution nucleophilic bimolecular	
TASF	Tris(dimethylamino)sulfonium trimethylsilyldifluoride	2f
THF	Tetrahydrofuran	
TMS	Trimethylsilyl	3a
TMSA	Trimethylsilylacetylene	3a
UV/Vis	Ultra Violet/Visible Light	
X	Polar head groups	3b
Y	Chain Length	

## Compound Numbering Scheme

Structure	No
 4-Phenylethyneaniline	1
 4-Hexyloxybenzaldehyde	2a
 4-Heptyloxybenzaldehyde	2b
 4-Octyloxybenzaldehyde	2c
 4-Nonyloxybenzaldehyde	2d
 4-Decyloxybenzaldehyde	2e
 Benzaldehyde	3a
 4-Hydroxybenzaldehyde	3b



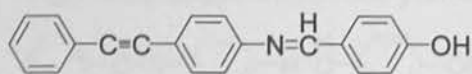
4-Bromobenzaldehyde

3c



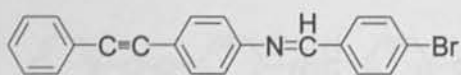
*N*-benzylidene-4-(phenylethynyl)aniline

4a



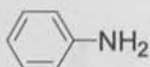
4-((4-phenylethynyl)phenyl)imino)methyl)phenol

4b



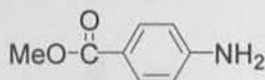
*N*-(4-bromobenzylidene)-4-(phenylethynyl)aniline

4c



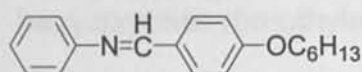
Phenylamine

5a



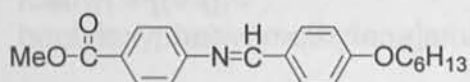
4-Amino methylbenzoate

5b



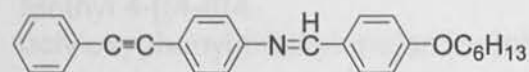
*N*-(4-hexyloxybenzylidene)aniline

6a



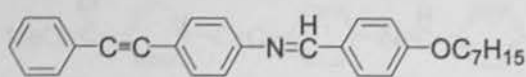
Methyl 4-[(4-hexyloxybenzylidene)amino]benzoate

6b



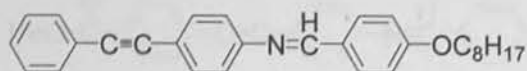
*N*-[(4-hexyloxyphenyl)methylene]-4-(phenylethynyl)aniline

7a



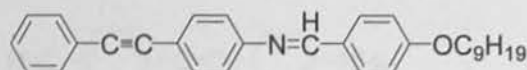
7b

*N*-[(4-heptyloxyphenyl)methylene]-4-(phenylethynyl)aniline



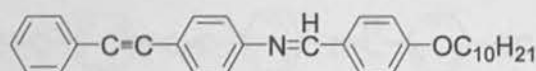
7c

*N*-[(4-octyloxyphenyl)methylene]-4-(phenylethynyl)aniline



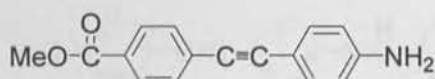
7d

*N*-[(4-nonyloxyphenyl)methylene]-4-(phenylethynyl)aniline



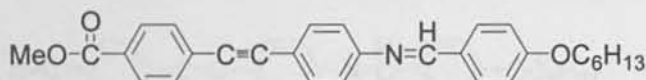
7e

*N*-[(4-decyloxyphenyl)methylene]-4-(phenylethynyl)aniline



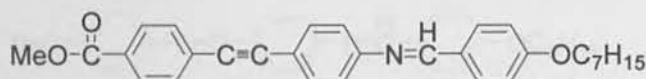
8

Methyl 4-[(4-aminophenyl)ethynyl]benzoate



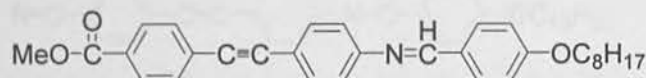
9a

Methyl 4-[(4-[(4-hexyloxyphenyl)methylene]amino}phenyl)ethynyl]benzoate



9b

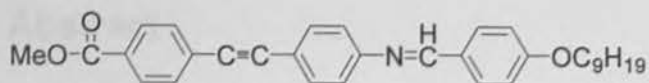
Methyl 4-[(4-[(4-heptyloxyphenyl)methylene]amino}phenyl)ethynyl]benzoate



9c

Methyl 4-[(4-[(4-octyloxyphenyl)methylene]amino}phenyl)ethynyl]benzoate





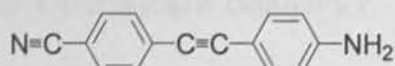
9d

Methyl 4-[(4-[(4-nonyloxyphenyl)methylene]amino)phenyl]ethynyl]benzoate



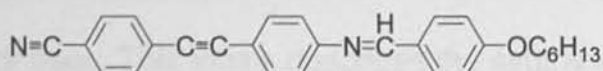
9e

Methyl 4-[(4-[(4-decyloxyphenyl)methylene]amino)phenyl]ethynyl]benzoate



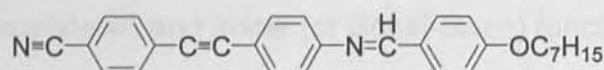
10

4-[(4-Aminophenyl)ethynyl]benzotrile



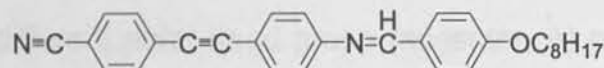
11a

4-[(4-[(4-Hexyloxyphenyl)methylene]amino)phenyl]ethynyl]benzotrile



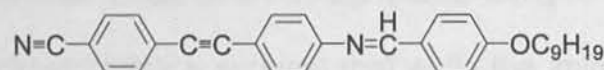
11b

4-[(4-[(4-Heptyloxyphenyl)methylene]amino)phenyl]ethynyl]benzotrile



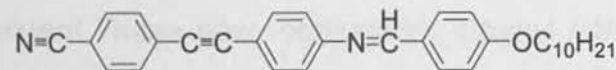
11c

4-[(4-[(4-Octyloxyphenyl)methylene]amino)phenyl]ethynyl]benzotrile



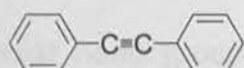
11d

4-[(4-[(4-Nonyloxyphenyl)methylene]amino)phenyl]ethynyl]benzotrile



11e

4-[(4-[(4-Decyloxyphenyl)methylene]amino)phenyl]ethynyl]benzotrile



12

Diphenylacetylene

## Abstract

To date conjugated ethynylated aromatic Schiff-Base systems are largely unexplored although the combination of two such well-known  $\pi$ -systems promises a wide range of electronic properties ranging from efficient electronic transmission to luminescent behaviour. The rigid linear nature of each group has led to the development of systems which exhibit liquid crystalline (LC) properties, and the combination of these motifs should be expected to lead to new materials with LC phases. This thesis describes the synthesis, molecular and electronic structure, as well as liquid crystalline behaviour, of a novel family of compounds featuring both acetylenic and imine (or Schiff-Base) functionalities.

Three series of ethynylated aromatic Schiff-Base systems were synthesised with a different polar head group (acceptor, A) namely; H, MeCO<sub>2</sub> and C $\equiv$ N and various chain length alkoxy (donor, D) tails, to give rise to compounds which feature an unique D-C<sub>6</sub>H<sub>4</sub>-CH=N-C<sub>6</sub>H<sub>4</sub>-C $\equiv$ C-C<sub>6</sub>H<sub>4</sub>-A substructure. Preliminary photophysical characteristics suggest that while the imine portion of the molecule dominates the electronic transitions the arylacetylene moiety must be involved to some extent. These new, conjugated ethynyl / Schiff-Base hybrid systems exhibit liquid crystalline properties at elevated temperatures. While all of the compounds examined have nematic phase, the compounds which feature longer alkyl tails or polar head groups also give rise to a Smectic A and/or Smectic B (hexatic B) phases.