

EFFECT OF SALT RATE ON THE PERFORMANCE OF  
MANGROVE PLANTS IN THE  
REGION OF MALAYSIA

ROSEMARY DUNN-JONES

FACULTY OF SCIENCE AND TECHNOLOGY  
UNIVERSITY OF MALAYA, SEREMBANG

2597

1100051095

LP 27 FST 1 2007



1100051095

Effect of shear rate on the performance of nanofiltration  
membrane in the removal of ammonium ion / Norhafizah Mat  
Junus.



PERPUSTAKAAN  
UNIVERSITI MALAYSIA TERENGGANU (UMT)  
21030 KUALA TERENGGANU

1100051095

1100051095		

Lihat sebelah

HAK MILIK  
PERPUSTAKAAN UMT

EFFECT OF SHEAR RATE ON THE PERFORMANCE OF NANOFILTRATION  
MEMBRANE IN THE REMOVAL OF AMMONIUM ION

By

Norhafizah binti Mat Junus

Research report submitted in partial fulfillment of  
the requirements for the degree of  
Bachelor of Technology (Environmental)

Department of Engineering Sciences  
Faculty of Science and Technology  
UNIVERSITY MALAYSIA TERENGGANU  
2007



**JABATAN SAINS KEJURUTERAAN  
FAKULTI SAINS DAN TEKNOLOGI  
UNIVERSITI MALAYSIA TERENGGANU**

**BORANG PENGAKUAN DAN PENGESAHAN LAPORAN  
PROJEK PENYELIDIKAN I DAN II**

Adalah ini diakui dan disahkan bahawa laporan penyelidikan bertajuk:

**Effect of Shear Rate on the Performance of Nanofiltration Membrane in the Removal of Ammonium Ion** oleh NORHAFIZAH BINTI MAT JUNUS No. Matrik UK7804 telah diperiksa dan semua pembetulan yang disarankan telah dilakukan. Laporan ini dikemukakan kepada Jabatan Sains Kejuruteraan sebagai memenuhi sebahagian daripada keperluan memperoleh Ijazah SARJANA MUDA TEKNOLOGI (ALAM SEKITAR), Fakulti Sains dan Teknologi, Universiti Malaysia Terengganu.

Disahkan oleh,

.....

Penyelia Utama

Nama: Encik Asmadi bin Ali @ Mahmud

Cop Rasmi:

**ASMADI ALI @ MAHMUD**  
Pensyarah  
Jabatan Sains Kejuruteraan  
Fakulti Sains dan Teknologi  
Universiti Malaysia Terengganu  
21030 Kuala Terengganu.

Tarikh: 24/5/17

.....

Penyelia Kedua (jika ada)

Nama: Dr. Nora'aini binti Ali

Cop Rasmi: DR. NORA'AINI BINTI ALI  
Ketua

Jabatan Sains Kejuruteraan  
Fakulti Sains dan Teknologi  
Universiti Malaysia Terengganu  
21030 Kuala Terengganu

Tarikh: 27/5/17

.....

Ketua Jabatan Sains Kejuruteraan

Nama: Dr. Nora'aini binti Ali

Cop Rasmi: DR. NORA'AINI BINTI ALI  
Ketua

Jabatan Sains Kejuruteraan  
Fakulti Sains dan Teknologi  
Universiti Malaysia Terengganu  
21030 Kuala Terengganu

Tarikh: 29/5/17

## ACKNOWLEDGEMENT

First of all I would like to thank to Allah S.W.T for blessing me and giving me the strength in completing my thesis.

I would like to express my warmest appreciation to my main supervisor and also as the coordinator for the final year project, Mr. Asmadi Ali, for all his advices, guidance and constructive criticisms throughout this study. Also, to my co-supervisor Dr. Nora'aini Ali, for her advices, supports and comments. Besides that, I would like to express my deepest appreciation to all the Department of Engineering Science staffs that had been giving me a hand.

I would like to express my special thanks to master students and my membrane's group members supervised under Mr. Asmadi Ali and Dr. Nora'aini Ali for helping me during this study. And not forgetting to all my friends and course mates.

Lastly, it is my pleasure to thank my parents and siblings for giving me mental and financial supports. Hope all of you have a successful and peaceful life.

## CONTENT

	<b>Page</b>
<b>TITLE PAGE</b>	i
<b>CONFIRMATION AND APPROVAL OF REPORT</b>	ii
<b>ACKNOWLEDGEMENT</b>	iii
<b>TABLE OF CONTENT</b>	iv
<b>LIST OF TABLES</b>	vii
<b>LIST OF FIGURES</b>	viii
<b>LIST OF ABBREVIATIONS</b>	x
<b>LIST OF APPENDICES</b>	xi
<b>ABSTRACT</b>	xii
<b>ABSTRAK</b>	xiii
<b>CHAPTER 1</b>	
<b>INTRODUCTION AND OBJECTIVES</b>	1
1.1 Problem Statement	3
1.2 Objectives of Study	6
1.3 Scopes of Study	6
<b>CHAPTER 2</b>	
<b>LITERATURE REVIEW</b>	7
2.1 Membrane	7
2.1.1 <i>Membrane Classification</i>	9
2.1.2 <i>Membrane Separation Process</i>	10

2.2	Nanofiltration and Applications	11
2.3	Aquaculture waste	13
	2.3.1 <i>Ammonium ion</i>	14
	2.3.2 <i>Environmental Impact and Health Risk</i>	15
2.4	Membrane Performances	16
2.5	Effect of Shear Rate on Membrane Structure and Performance	17
2.6	Membrane Characterization	19
	2.6.1 <i>Steric Hindrance Pore (SHP) Model</i>	22
	2.6.2 <i>Teorell-Meyer-Sievers model</i>	24
<b>CHAPTER 3</b>	<b>METHODOLOGY</b>	<b>25</b>
3.1	Materials	25
	3.1.1 <i>Polymer (Polyethersulfone)</i>	25
	3.1.2 <i>Solvent (N-Methyl-2-Pyrrolidone)</i>	26
3.2	Dope Solution Preparation	26
3.3	Flat-sheet Membrane Fabrication	28
3.4	Scanning Electron Microscopy (SEM)	30
3.5	Membrane Performance Measurement	31
3.6	Analysis Technique for Ammonium Ion	34
	3.6.1 <i>Calibration Curve</i>	35
3.7	Determination of Pore Radius on Membrane Surface	37
3.8	Determination of Zeta Potential Using Electro Kinetics Analyzer (EKA)	37

<b>CHAPTER 4</b>	<b>RESULTS AND DISCUSSIONS</b>	39
4.1	Membrane Characterization	39
4.1.1	<i>Effect of Shear Rate on the Pure Water Flux Measurement</i>	40
4.1.2	<i>Effect of Shear Rate on Membrane Separation Performance</i>	42
4.1.3	<i>Theoretical Approach using SHP Model</i>	46
4.1.4	<i>Surface Charge Prediction using Electro Kinetics Analyzer (EKA)</i>	48
4.2	Membrane Performances	50
4.2.1	<i>Effect of Shear Rate on the Removal of Ammonium Ion</i>	51
4.2.2	<i>The Optimum Shear Rate with the Best Membrane Performance</i>	55
4.3	Effect of Shear Rate on Membrane Morphology	57
<b>CHAPTER 5</b>	<b>CONCLUSIONS AND RECOMMENDATIONS</b>	61
5.1	Conclusions	61
5.2	Recommendations	63
<b>REFERENCES</b>		65
<b>APPENDICES</b>		70
<b>VITAE</b>		89



## LIST OF TABLES

<b>No.</b>	<b>Tittle</b>	<b>Page</b>
2.1	Malaysia's standards water quality for ammonium-nitrogen	15
4.1	The permeability coefficient of membranes at different shear rates	42
4.2	Rejection and Flux of NaCl (0.01M) at various shear rate	45
4.3	Ammonium Ion rejection at different shear rate (%)	51
4.4	The productivity and selectivity of PES membrane in rejecting ammonium ion at different pressure	54
4.5	Relationship between permeability coefficients, ammonium ion flux and rejection at various shear rates	56

## LIST OF FIGURES

No.	Title	Page
2.1	Membrane separation process	8
3.1	Apparatus for preparation of casting solution	27
3.2	Semi-automated Electrically Controlled Casting Machine	28
3.3	Coagulation bath	29
3.4	Scanning Electron microscopy (SEM)	30
3.5	Sterlich Dead-end Stirred Cell	31
3.6	Schematic diagram of zeta potential measurement	38
4.1	Permeability response regression lines and equation for membrane of shear rate $278.88\text{s}^{-1}$ , $185.92\text{s}^{-1}$ , $139.44\text{s}^{-1}$ , $111.55\text{s}^{-1}$ and $92.96\text{s}^{-1}$	41
4.2	Effect of shear rate on NaCl (0.01M) based on (a) percentage of rejection and (b) flux	43
4.3	Relationship between the flux and rejection rate of NaCl solutions (0.01M) versus shear rates	45
4.4	Pore radius and percentage of rejection vs. shear rates	47
4.5	Pore radius and fluxes vs. shear rates	47
4.6	Rejection and $\zeta$ -potentials as function of various shear rates	49
4.7	Flux and $\zeta$ -potentials as function of various shear rates	49
4.8	Ammonium ion calibration curve, absorbance versus concentration	50
4.9	Ammonium ion rejection at different shear rates of membrane	52

4.10	The fluxes for ammonium ion rejection at different shear rate of membrane	52
4.11	Relationship between the rejection and the flux value according to the various shear rates	55
4.12	Cross-section of PES membranes at different shear rate: (a) $278.88\text{s}^{-1}$ , (b) $185.92\text{s}^{-1}$ , (c) $139.44\text{s}^{-1}$ , (d) $111.55\text{s}^{-1}$ , (e) $92.96\text{s}^{-1}$	60

## LIST OF ABBREVIATIONS

### Abbreviations

APHA	American Public Health Association
AWWA	American Water Works Association
EKA	Electro Kinetics Analyzer
INOS	Institute Oceanography
MF	Microfiltration
MW	Molecular Weight
NF	Nanofiltration
NMP	N-methyl-2-pyrrolidone
PES	Polyethersulfone
RO	Reverse Osmosis
SEM	Scanning Electron Microscopy
SHP	Steric Hindrance Pore model
UF	Ultrafiltration

## APPENDICES

<b>Appendix</b>		<b>Page</b>
A	Gantt chart	70
B	Phenol-hypochlorite reagent	71
C	Ammonia Readings (Absorbance)	72
D	Pure water flux and rejection data at various shear rate	73
E	Data for theoretical approach using SHP Model	85
F	Data for theoretical approach using EKA	89

## ABSTRACT

Excessive ammonia in aquaculture wastewater may cause many environmental problems to the receiving water. Such adverse effects of ammonium have promoted developing various techniques for its removal. Membrane separation processes are largely developed as an alternative to conventional wastewater treatment. In order to fabricate a high performance of nanofiltration membrane, the effect of shear rate on the membrane performance for aquaculture wastewater treatment has been studied. Membrane casting solution was prepared in binary system which consists of Polyethersulfone (PES) as polymer and N-methyl-2-pyrrolidone (NMP) as solvent with formulation 16/84 (PES/NMP). The flat sheet membranes were produced by a dry/wet phase inversion technique using a semi-automated electrically controlled casting machine varying casting speed 10s, 15s, 20s, 25s and 30s. Varying the casting speed has produced membranes with shear rate  $278.88\text{s}^{-1}$ ,  $185.92\text{s}^{-1}$ ,  $139.44\text{s}^{-1}$ ,  $111.55\text{s}^{-1}$ , and  $96.92\text{s}^{-1}$ . The pore radius and charge of the membrane were predicted using SHP model and EKA. It was found that the pore size was decreased and the membrane charge implied the highest  $\xi$ -potentials (most negative) with increasing shear rate. The membranes performances were study based on percentage of rejection and flux of sodium chloride (NaCl) solution and ammonium solution. The highest rejection and flux for NaCl was found to be about 57% and  $0.150\text{ L/m}^2\cdot\text{h}$  respectively at shear rate  $278.88\text{s}^{-1}$ . The same trend was also observed for ammonium ion removal where the membrane at shear rate  $278.88\text{s}^{-1}$  removes greater than 70% of ammonium ion with  $324.44\text{ L/m}^2\cdot\text{h}$  flux. Due to the smallest pore radius, most negative surface charge and long finger-like structure, membrane with shear rate  $278.88\text{s}^{-1}$  produced the highest rejection rate and flux rate. These findings suggest that the best shear rate is  $278.88\text{s}^{-1}$ .

## ABSTRAK

Pelbagai masalah alam sekitar boleh terjadi akibat kandungan ammonia yang berlebihan dalam air sisa akuakultur. Untuk mengatasi masalah tersebut, pelbagai teknik telah dibangunkan. Proses pemisahan menggunakan membran telah berkembang secara meluas sebagai alternatif baru dalam rawatan air sisa. Kesan kadar ricih ke atas prestasi membran telah dikaji dalam proses untuk menghasilkan membran penuras nano berprestasi tinggi. Larutan membran sistem binari yang terdiri daripada polietersulfon (PES) sebagai polimer dan n-metil-2-pirrolidon (NMP) sebagai pelarut dengan formulasi 16/84 (PES/NMP) telah dihasilkan. Kepingan nipis membran dihasilkan melalui kaedah pembalikan fasa kering/basah dengan menggunakan mesin acuan elektrik separa automatik dengan kelajuan 10s, 15s, 20s, 25s, dan 30s. Kelajuan mesin acuan tersebut menghasilkan membran berkadar ricih  $278.88s^{-1}$ ,  $185.92s^{-1}$ ,  $139.44s^{-1}$ ,  $111.55s^{-1}$ , dan  $96.92s^{-1}$ . Jejari liang dan cas pada permukaan membran telah ditentukan menggunakan model SHP dan EKA. Telah dikenalpasti bahawa saiz liang semakin mengecil dan membran semakin bercas negatif dengan pertambahan kadar ricih. Prestasi membran dikaji berdasarkan peratus penyingkiran dan fluk larutan NaCl dan larutan ammonia. Pada kadar ricih  $278.88s^{-1}$ , peratus penyingkiran tertinggi NaCl adalah sebanyak 57% dengan fluk  $0.150 L/m^2.h$ . Paten yang sama didapati pada penyingkiran ion ammonia dimana lebih 70% ion ammonia disingkirkan dengan fluk  $324.44 L/m^2.h$  pada kadar ricih  $278.88s^{-1}$ . Disebabkan saiz liang paling kecil, permukaan membran yang semakin bercas dan struktur seakan jejari yang panjang pada lapisan aktif membran telah menghasilkan peratusan penyingkiran dan flux yang tinggi pada kadar ricih  $278.88s^{-1}$ . Hasil kajian tersebut menunjukkan kadar ricih terbaik adalah  $278.88s^{-1}$ .